Periods, Groups, Blocks and Valence Electrons (Worksheet 2)

Name _____

Fill in the blank cells in the tables.

Period	Group	Block	Element Name	Valence Electrons	Noble Gas Notation	Metal/non-metal or metalloid
4	13	p	Gallium	3	[Ar] $3d^{10}4s^{2}4p^{1}$	metal
2	2					
					[Ne] $3s^{2}3p^{3}$	
			Nitrogen			
3				1		
	18				[Ne}$3s^{2}3p^{6}$	
5	17					
					[Ar] $3d^{2}4s^{2}$	
2				7		
					[He] $2s^{1}$	
4	8					
					[Kr] $5s^{2}$	

Number of shells containing electrons	Number of electrons in the outermost shell (valence electrons)	Element Name	Period	Group
4	6	Selenium	4	16
			2	14
3				2
	8		3	
		Sulfur		
2	2			
3				17
	1		3	
		Calcium		
4				12
		Hydrogen		

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Name KEY

Fill in the blank cells in the tables.

Period	Group	Block	Element Name	Valence Electrons	Noble Gas Notation	Metal/non-metal or metalloid
4	13	p	Gallium	3	[Ar] $3d^{10}4s^2 4p^1$	metal
2	2	S	Beryllium	2	[He] $2s^2$	metal
3	15	P	Phosphorus	5	[Ne] $3s^2 3p^3$	non-metal
2	15	P	Nitrogen	5	[He] $2s^2 2p^3$	non-metal
3	1	S	Sodium	1	[Ne] $3s^1$	metal
3	18	P	Argon	8	[Ne] $3s^2 3p^6$	non-metal
5	17	P	Iodine	7	[Kr] $5s^2 4d^{10} 5p^5$	non-metal
4	4	d	Titanium	2	[Ar] $3d^2 4s^2$	metal
2	17	P	Fluorine	7	[He] $2s^2 2p^5$	non-metal
2	1	S	Lithium	1	[He] $2s^1$	metal
4	8	d	Iron	2	[Ar] $4s^2 3d^6$	metal
5	2	S	Strontium	2	[Kr] $5s^2$	metal

Number of shells containing electrons	Number of electrons in the outermost shell (valence electrons)	Element Name	Period	Group
4	6	Selenium	4	16
2	4	Carbon	2	14
3	2	Magnesium	3	2
3	8	Argon	3	18
3	6	Sulfur	3	16
2	2	Beryllium	2	2
3	7	chlorine	3	17
3	1	Sodium	3	1
4	2	Calcium	4	2
4	2	Zinc	4	12
1	1	Hydrogen	1	1