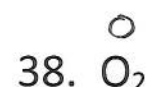
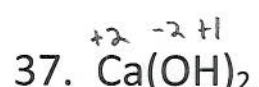
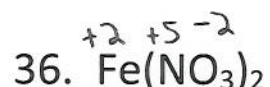
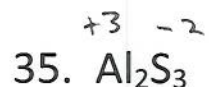
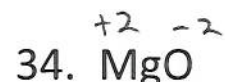
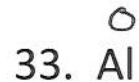
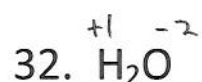
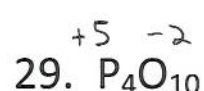
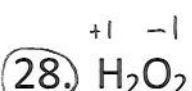
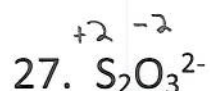
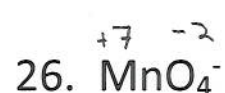
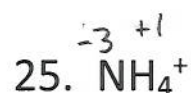
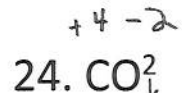
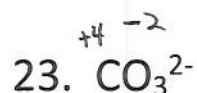
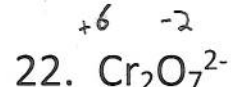
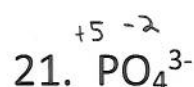
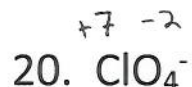
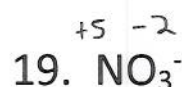
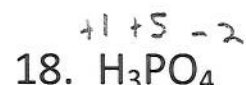
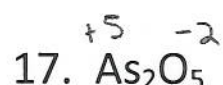
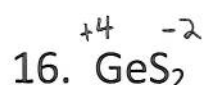
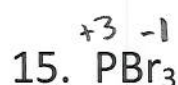
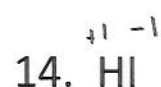
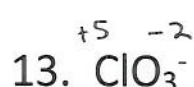
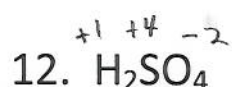
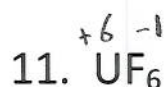
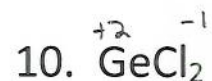
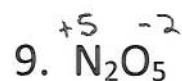
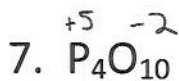
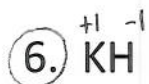
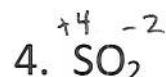
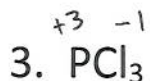
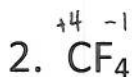
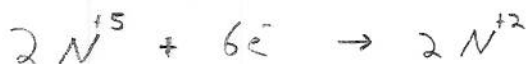
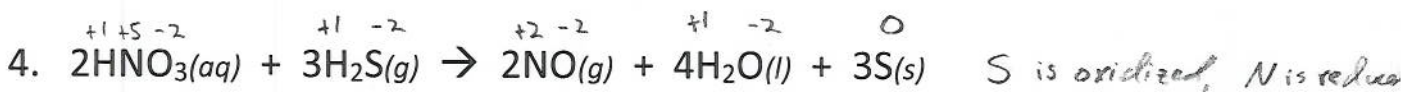
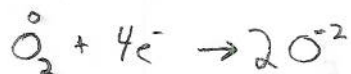
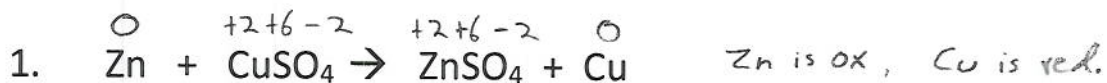


Name _____ Oxidation numbers

Assign oxidation numbers to each atom.



Assign oxidation numbers to each atom and identify what is oxidized and what is reduced.



5. What is the oxidation number of the Xe atom in each of the following:

